



Journées Chevreul
Maison Alfort, France
Juin, 5th 2012



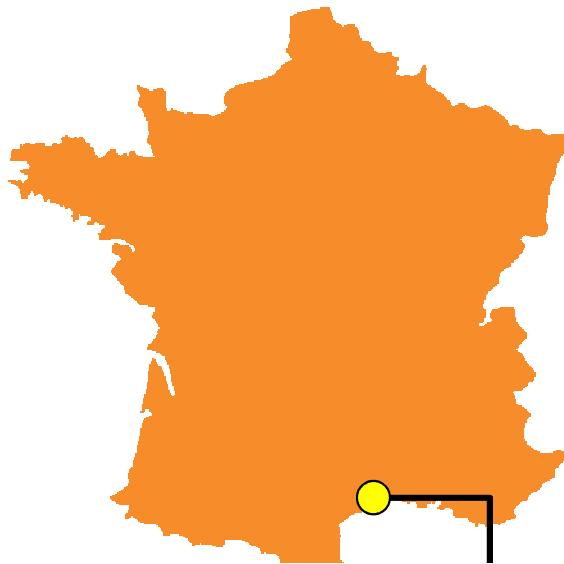
Synthesis of bio-based building blocks from vegetable oils: toward platform chemicals

Dr Sylvain Caillol

*Pr Bernard Boutevin, Dr Rémi
Auvergne*



CHEMISTRY IN MONTPELLIER, FRANCE

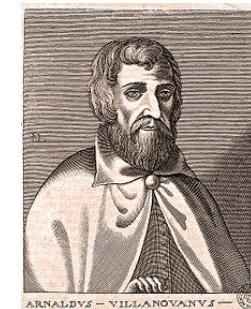


MONTELLIER

ChemSuD
Montpellier

- Arnaud De Villeneuve, 13th century Alchemist – acids and distillations...
- 1st Chemistry Institute created in 1889
- 2012 : 700 chemists and 1700 students
- Sustainable Chemistry Chair

<http://www.chemsud.fr>

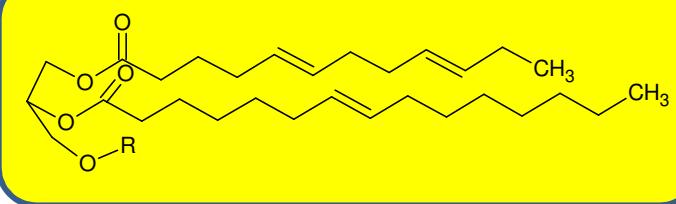


IAM Team : head Pr JJ Robin
Design and Architecture
for Macromolecules
60 pers.

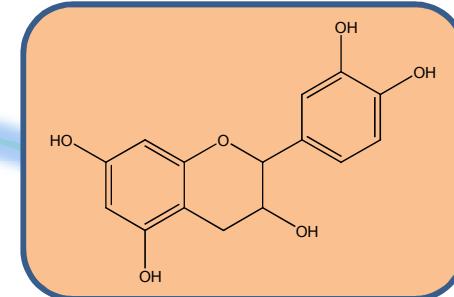
➤ Polymer team of the
Chemistry Institute **ICGM**



- Non harmful biobased intermediates for biobased polymers

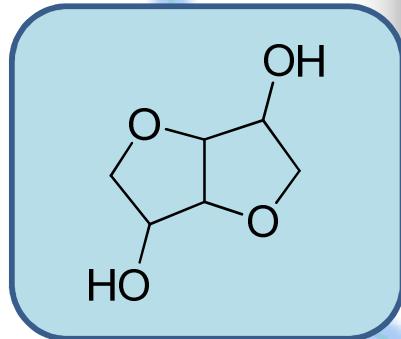


Vegetable oils

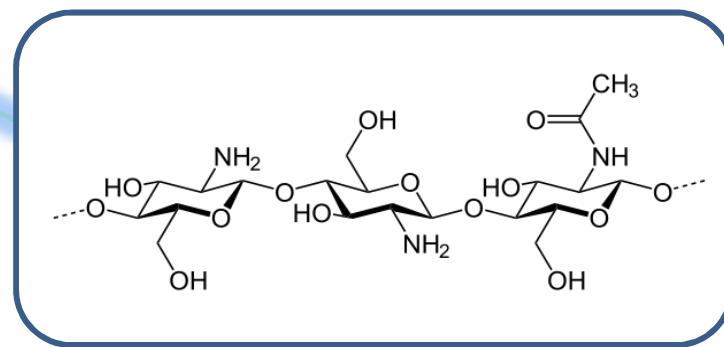


Polyphenols

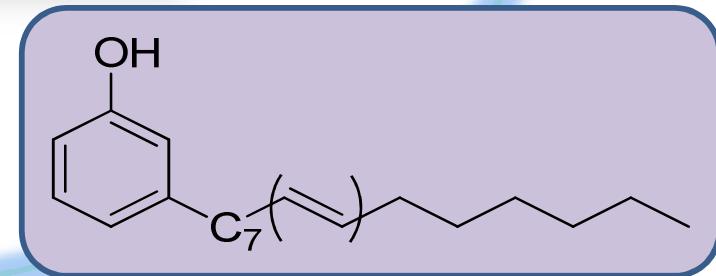
From molecular synthesis to functional materials : PU, epoxy, vinyl esters...



Isosorbide

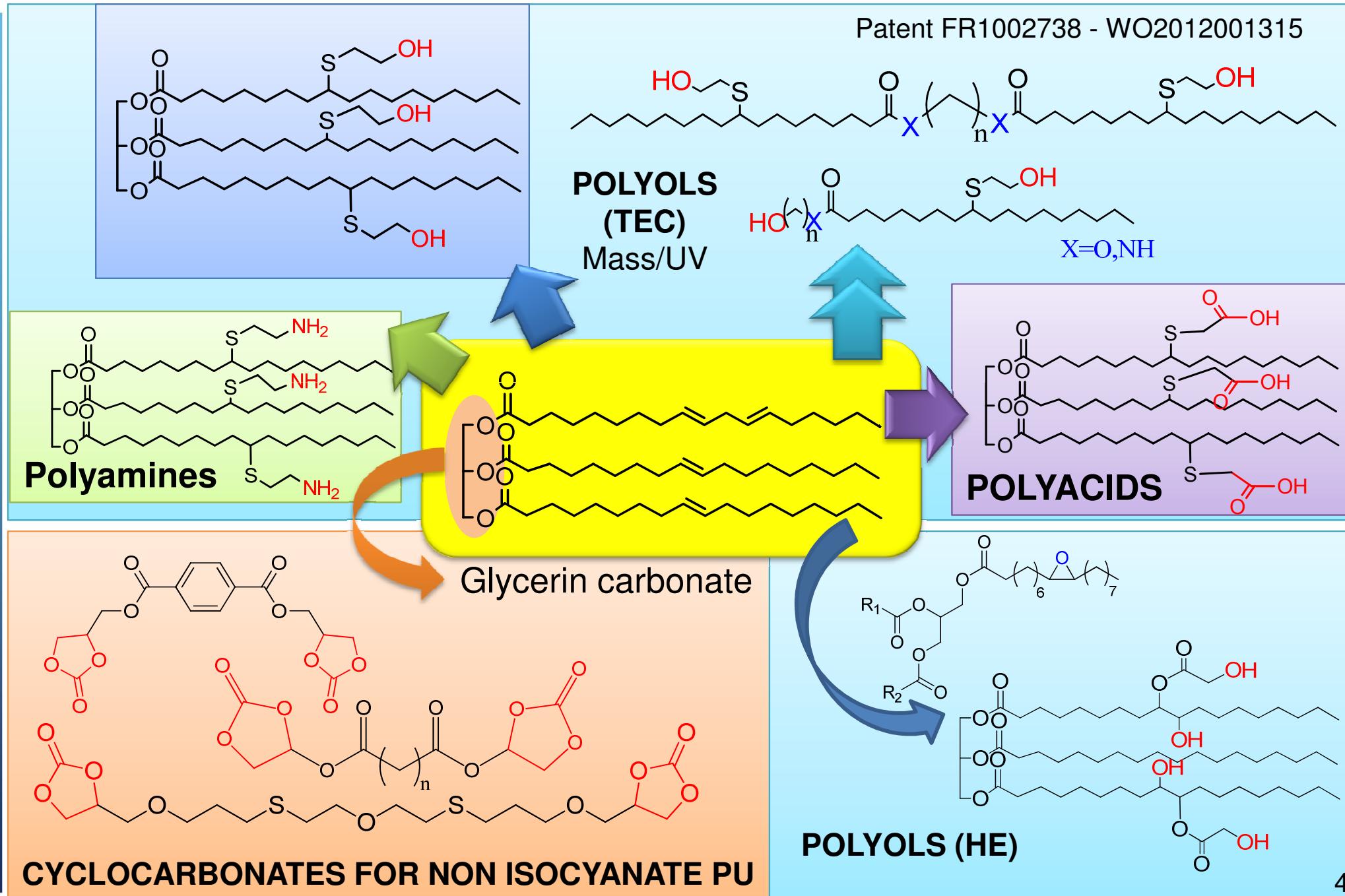


Polysaccharides



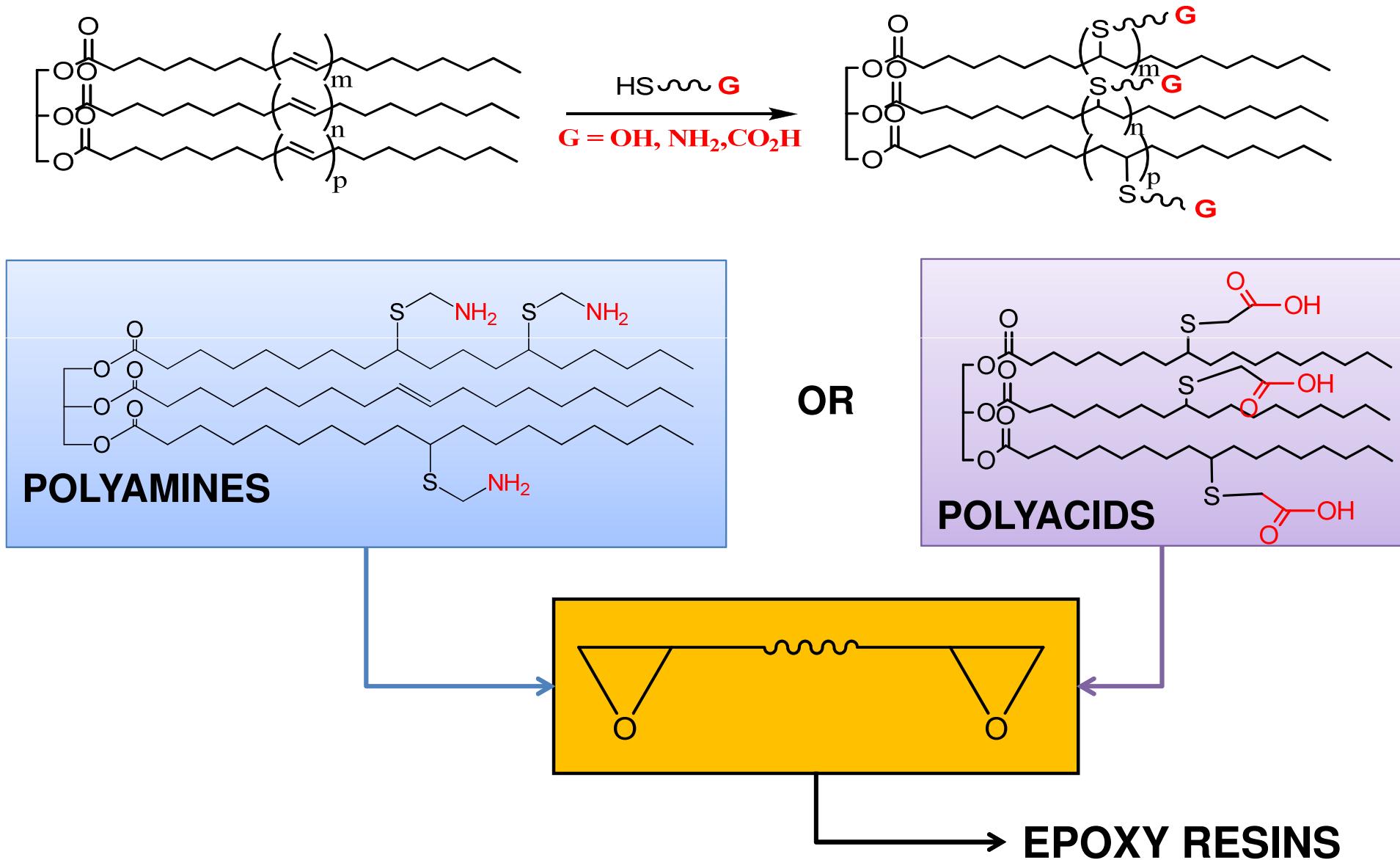
Cardanol

Vegetable oils: a platform molecule for new biobased intermediates

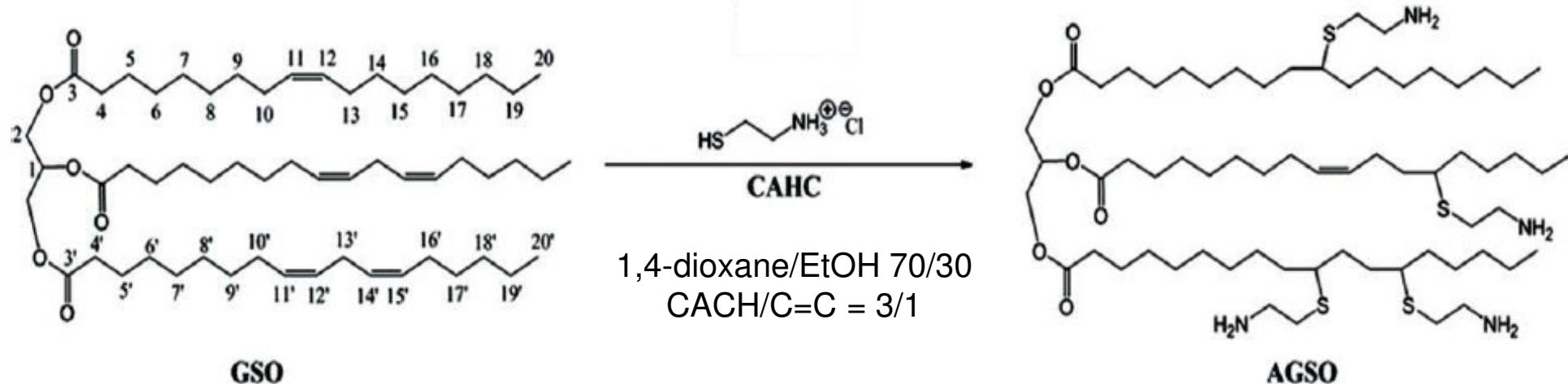


Part 1: Thiol-Ene Coupling Strategy

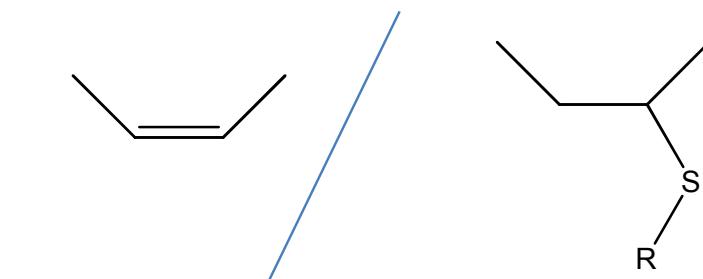
- Polyfunctional reactants from vegetable oil by Thiol-Ene Coupling



Grapeseed oil functionalized with cysteamine by TEC under UV

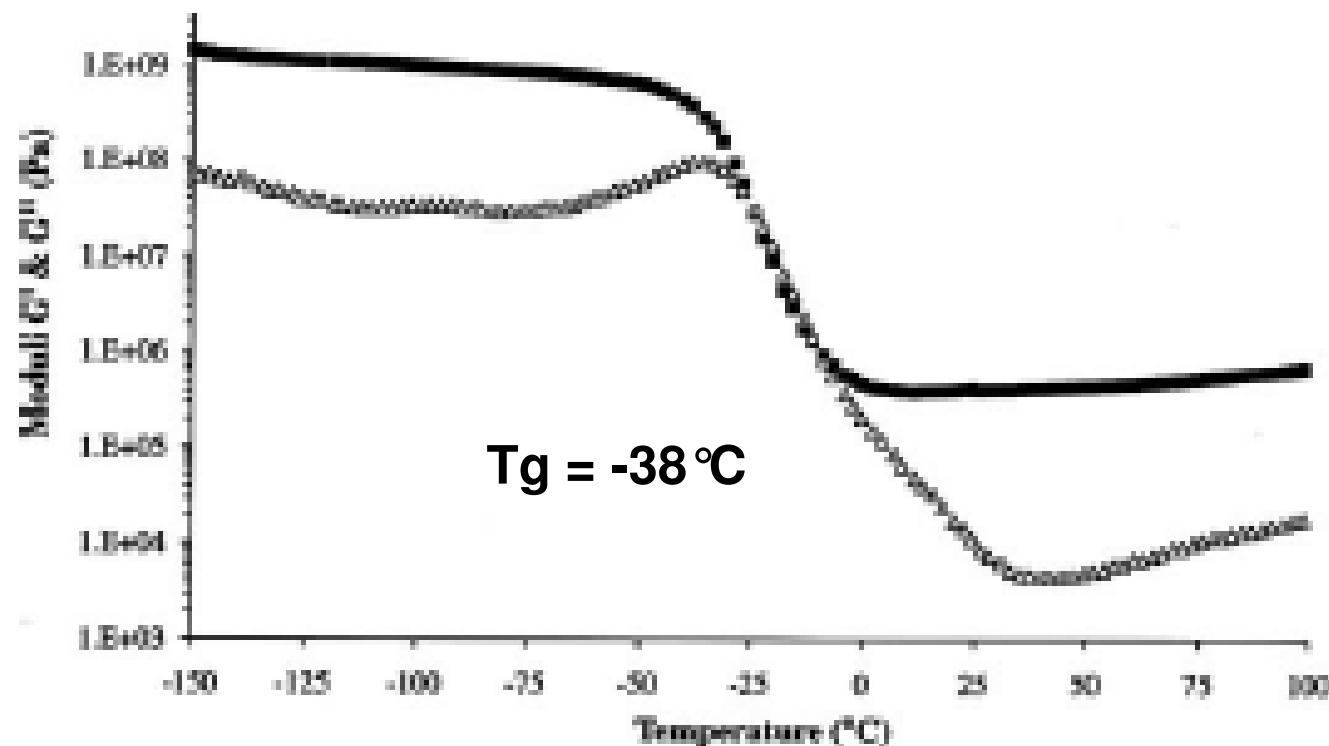
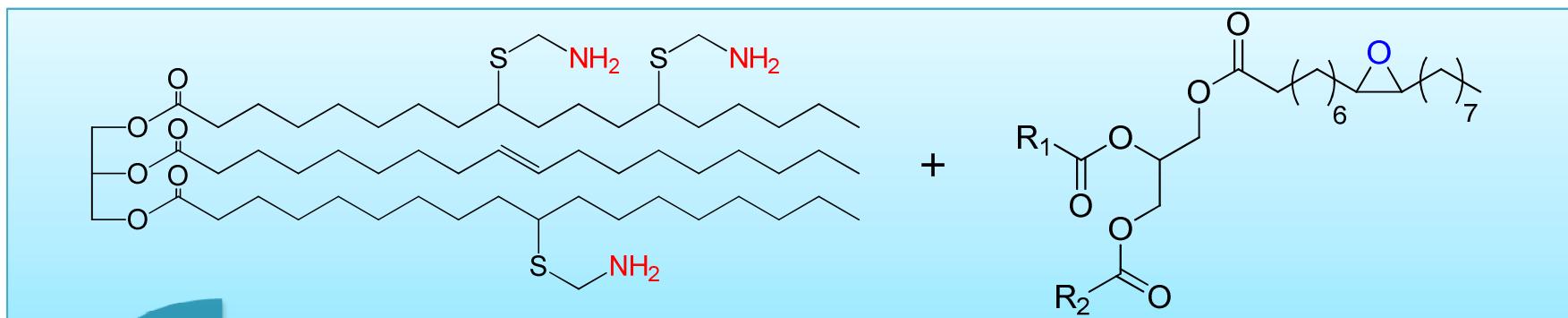


DMPA (eq)	<i>t</i> (h)	ρ (%) ^a / ρ (%) ^b
0.01	8	41/22
0.01	8	17/20
0.1	8	100/87
0.01	24	25/23
-	24	23/22
-	72	62/65
0.1	8	86/66



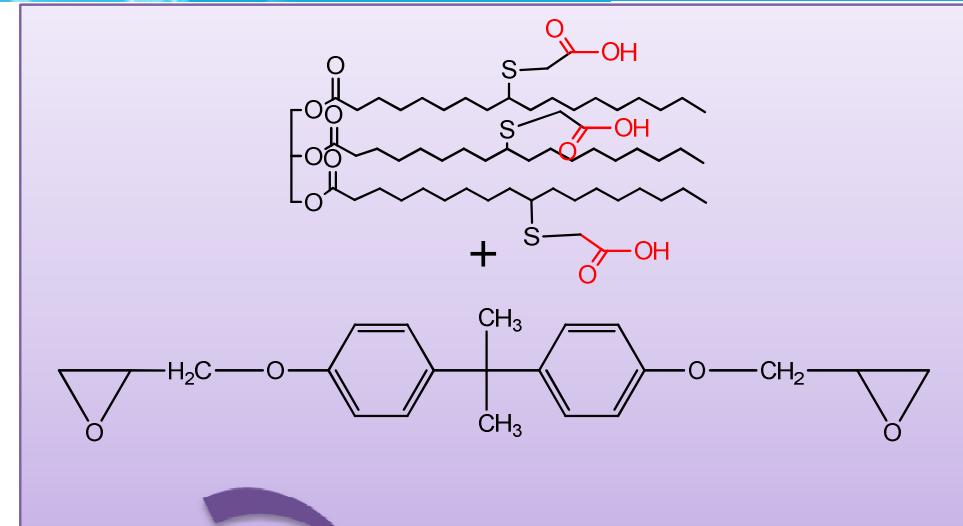
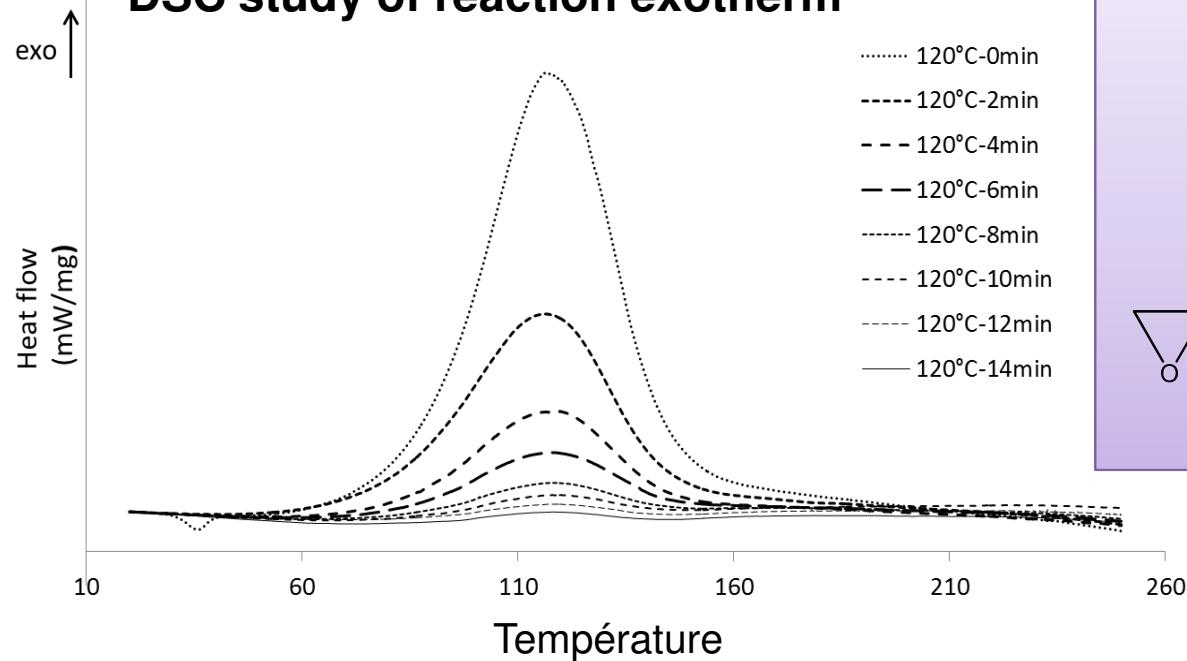
DMPA : 2,2-dimethoxy-2-phenylacetophenone

Polyamine and epoxidized oil - RT 24h

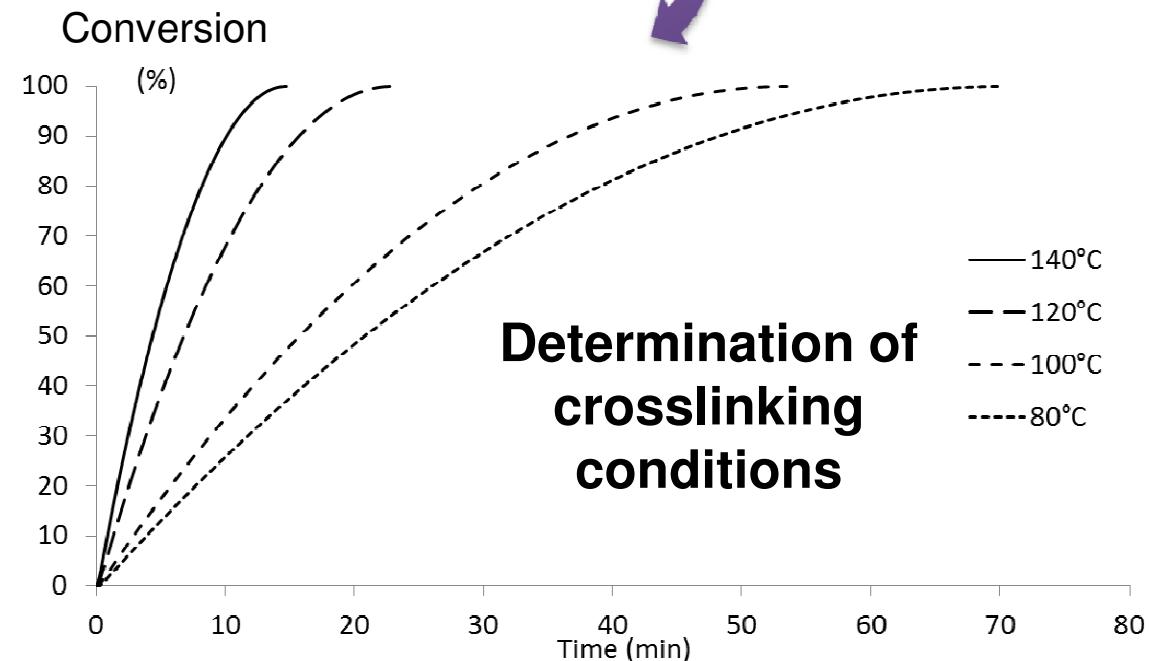


Thermomechanical analysis of cured AGSO-ELO mixture with G' (■) and G'' (Δ)

DSC study of reaction exotherm



Tg (°C)	TGA (°C) under N ₂ /air	
-12	348	340



Polyurethanes 6th polymer in the world - 14Mt/y various applications



➡ Issues

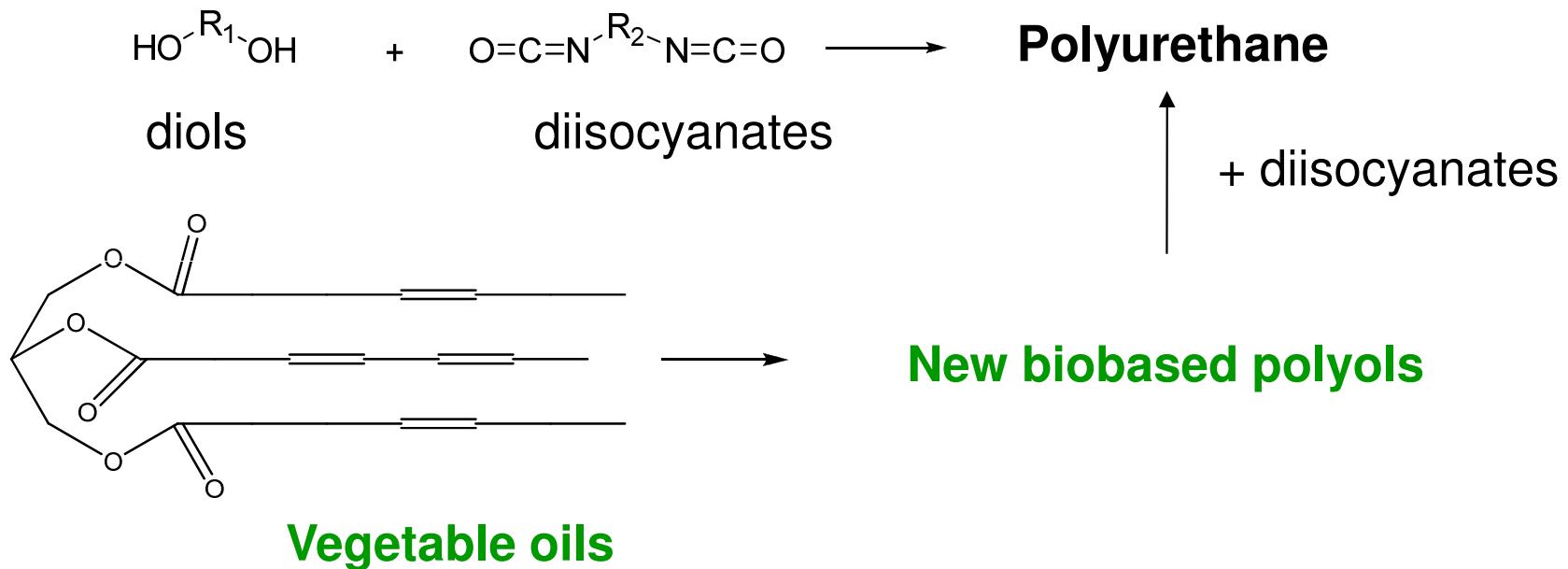
- **Fossil resources** reactants
- **Harmful** reactants
- Isocyanates - TDI and MDI : **CMR**
- Regulations
 - REACH: **MDI** in annex XVII

➡ Targets

- Use of **vegetable oils** # 20Mt used in chemistry
- Synthesize **biobased polyols** (70% w/w PU)
 - **Polyethers** 1st group of polyols for PU
 - **Polyesters** 2nd group of polyols for PU
- **PU without isocyanates: Non Isocyanate PU**

Objectives

- Synthesis of biobased polyurethanes materials from vegetable oils

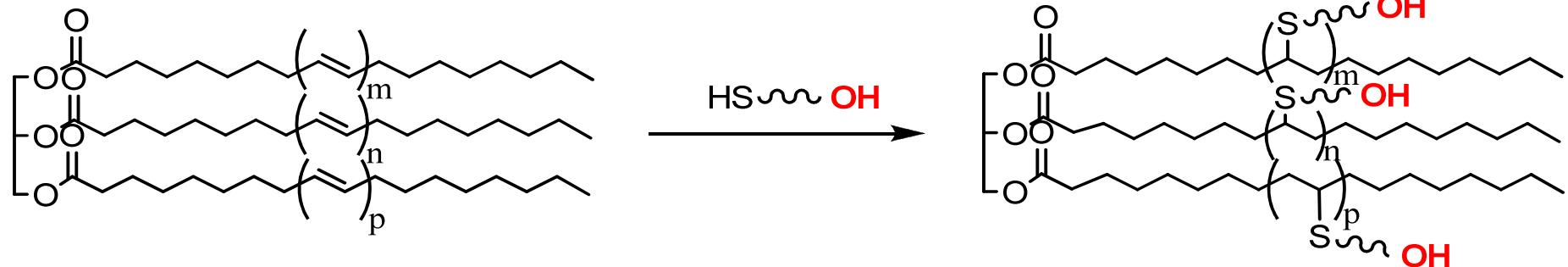


Method

- Direct hydroxyl functionalization of vegetable oils and derivatives by TEC

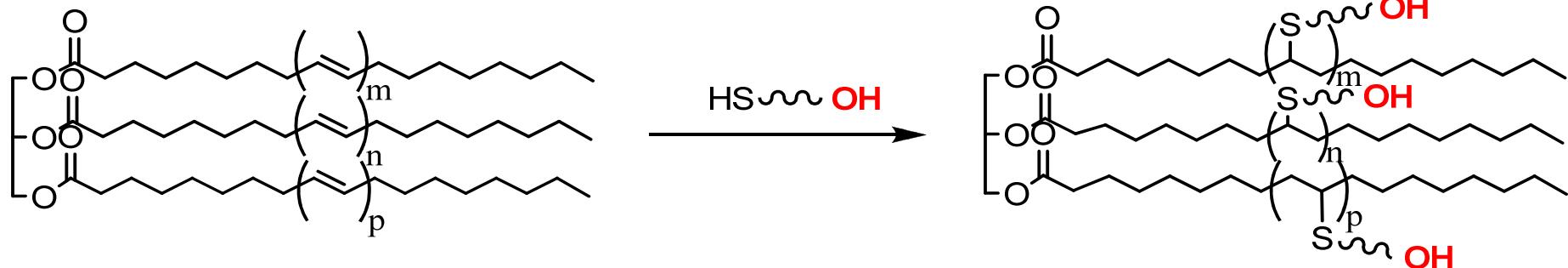
- Reactants: rapeseed oils and derivatives

- Route 1 : polyol from vegetable oil*

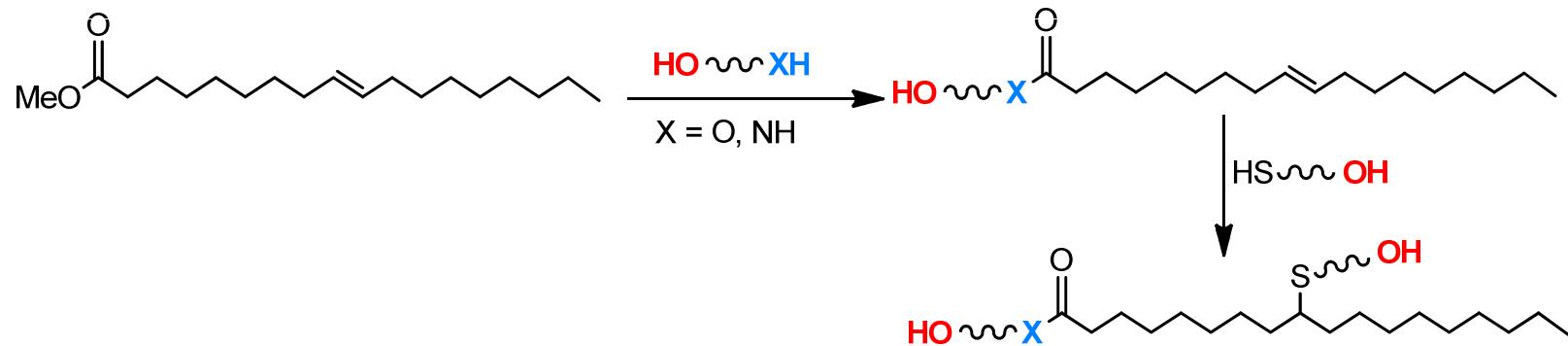


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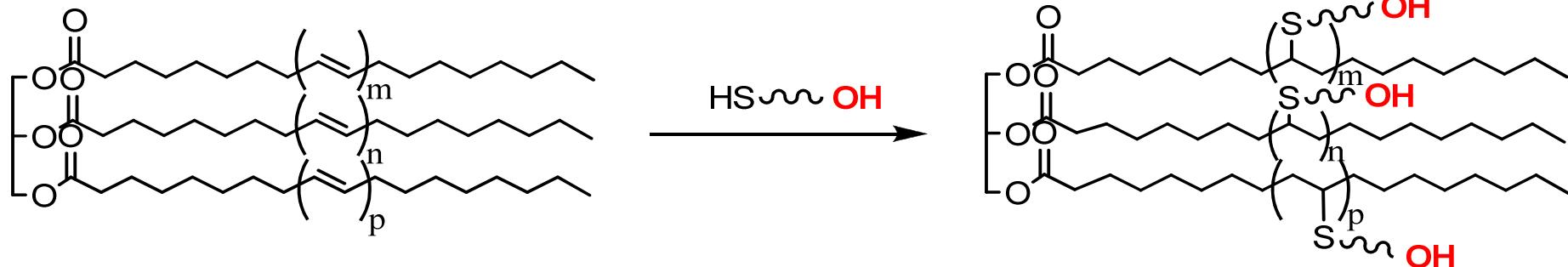


- Route 2 : diol from FAME*

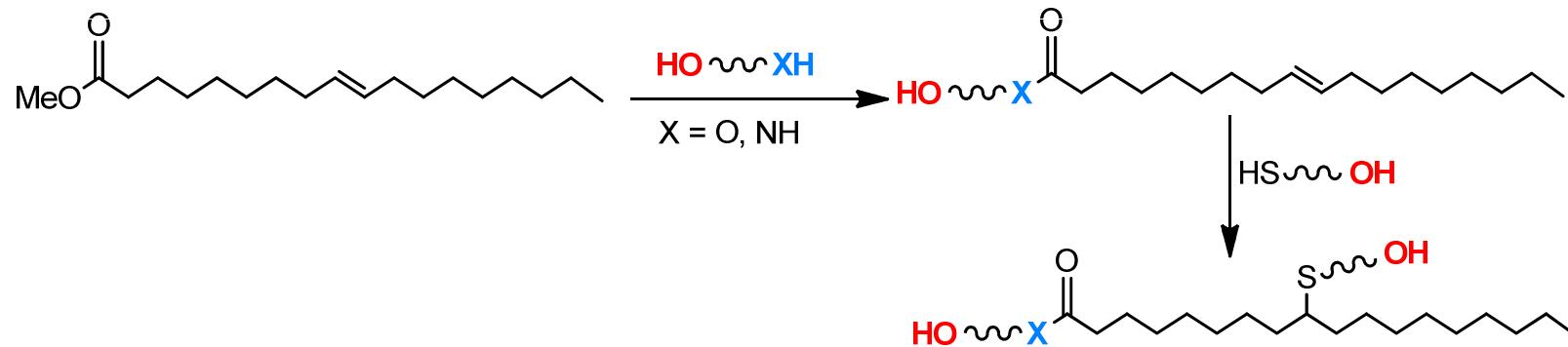


- Reactants: rapeseed oils and oleic acid or oleic methyl ester

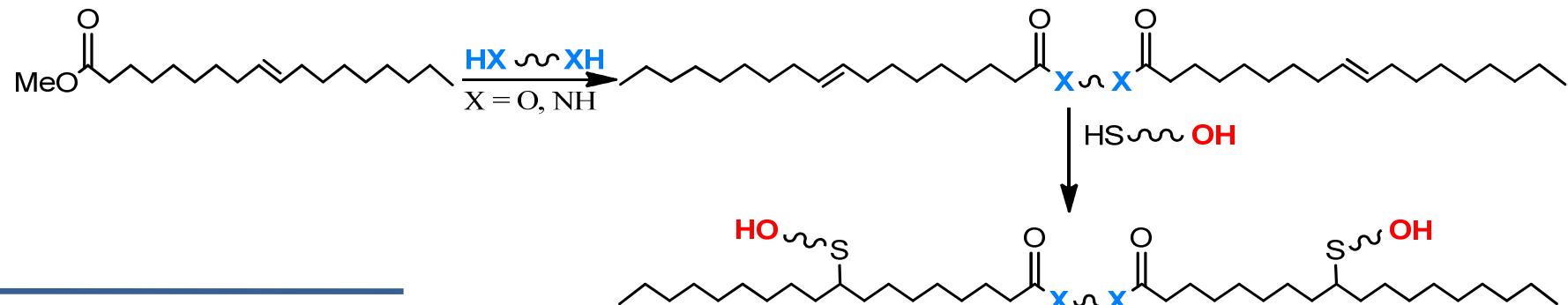
- Route 1 : polyol from vegetable oil*



- Route 2 : diol from FAME*

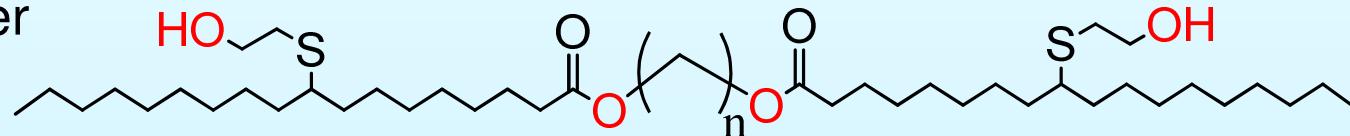


- Route 3 : diol dimers from FAME*

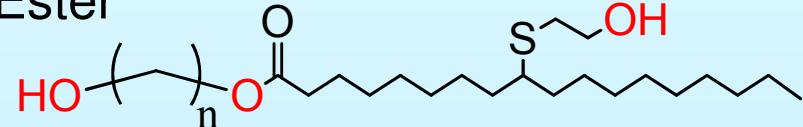


New biobased polyols from vegetable oils by TEC

Diester



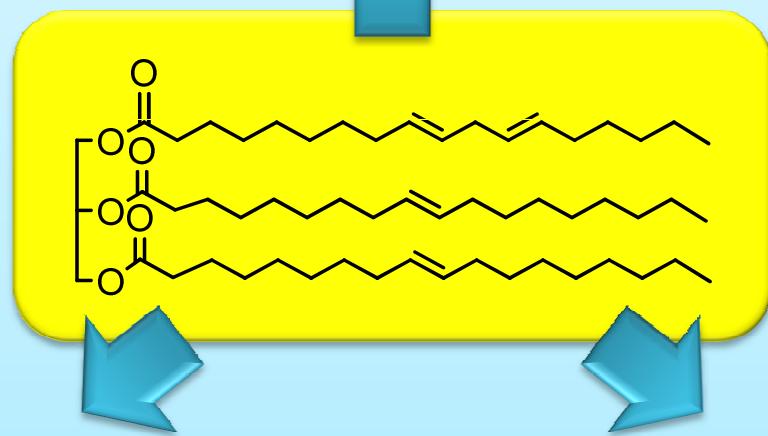
Ester



$n = 2, 4, 6$

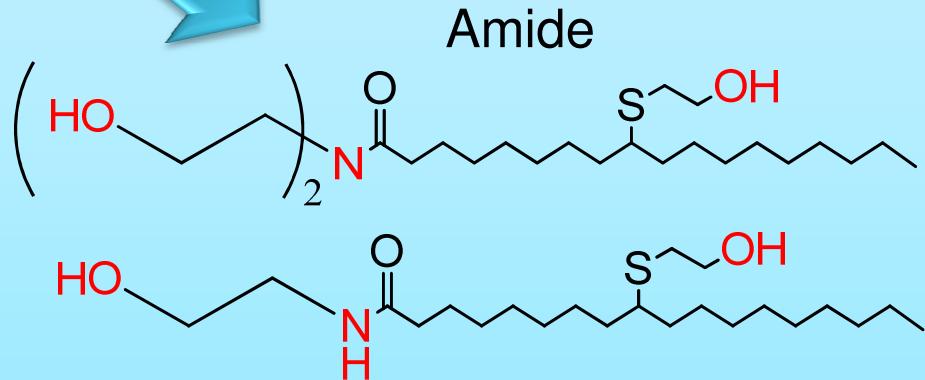
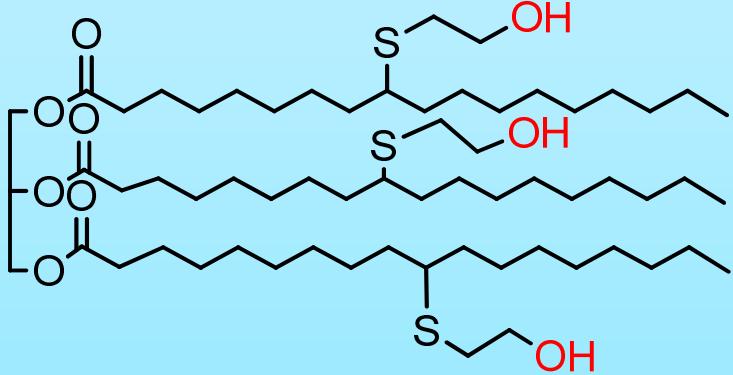


**Photoinitiation
without photoinitiator**
6h - UV, thiol/ene 3/1

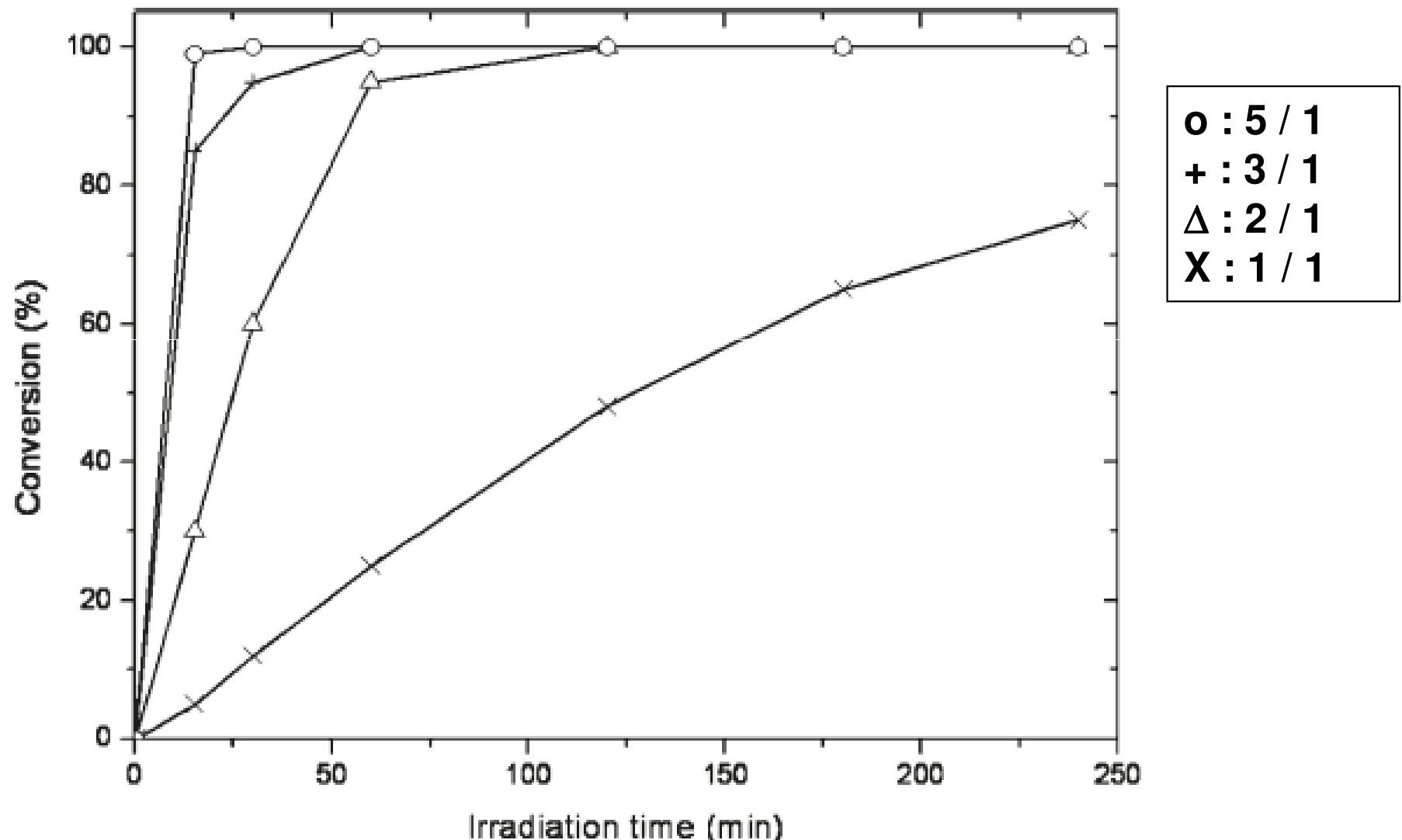


Thermal
AIBN
80 °C, 8h, thiol/ene 3/1

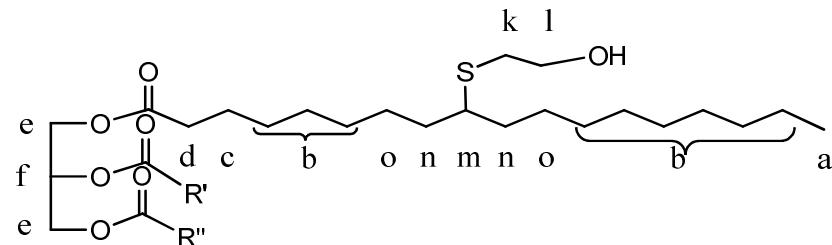
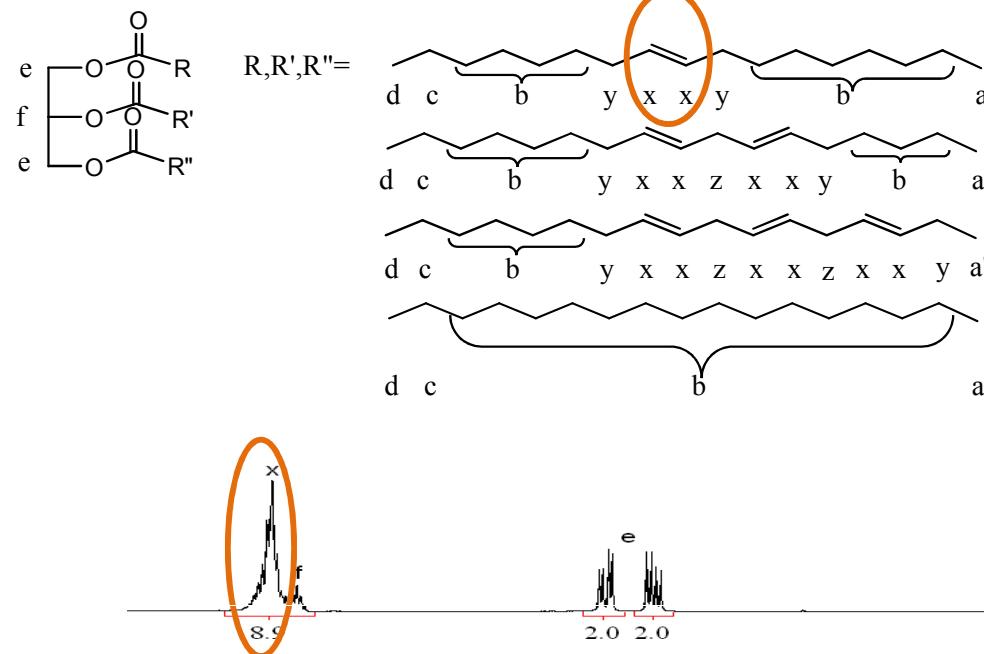
Oil



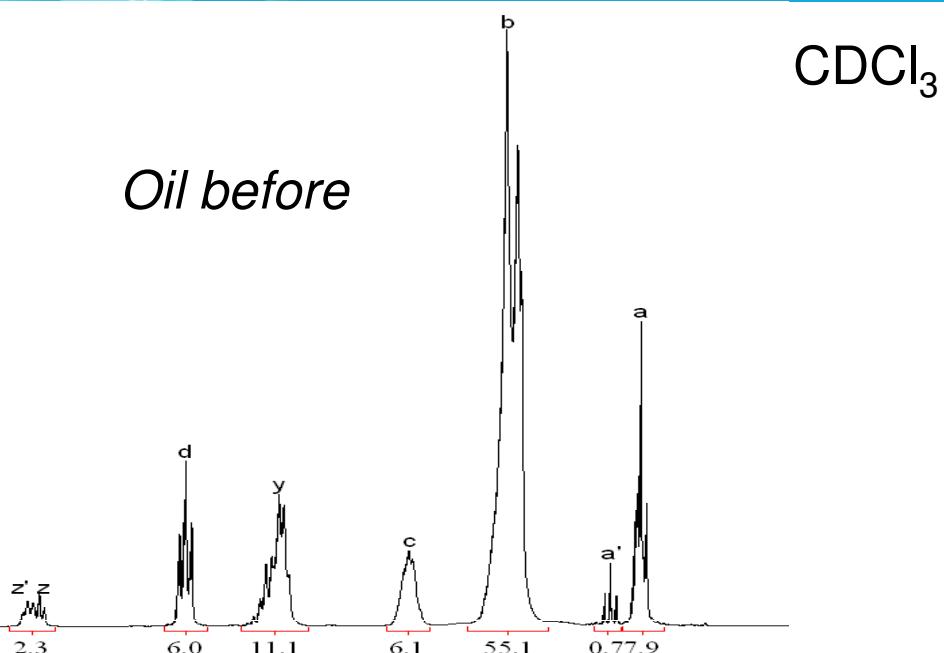
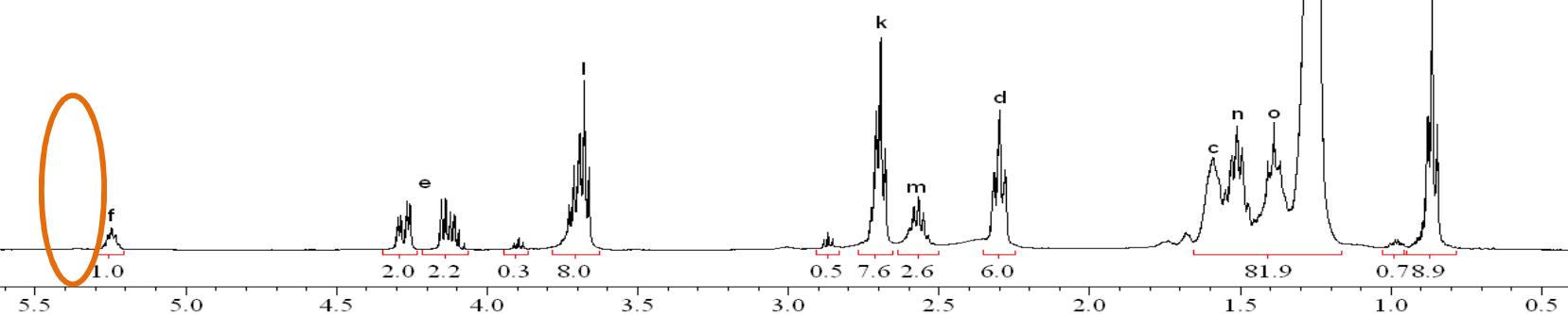
2-mercaptoproethanol/oleic acid



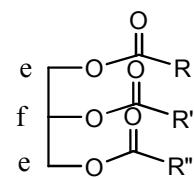
Polyols : NMR ^1H analysis



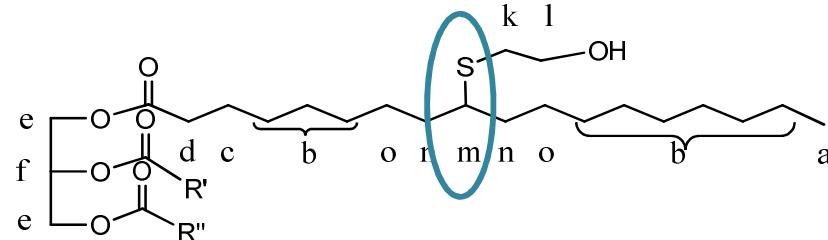
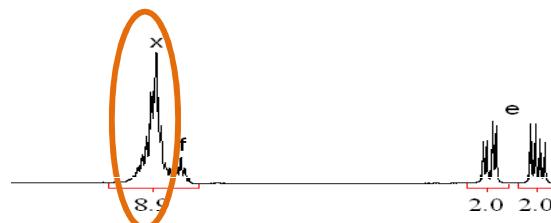
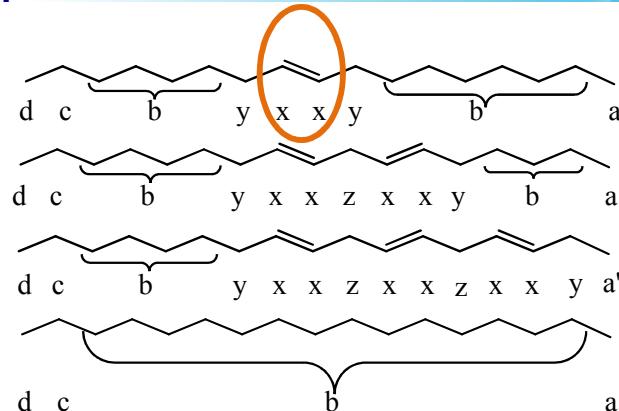
Total conversion of C=C



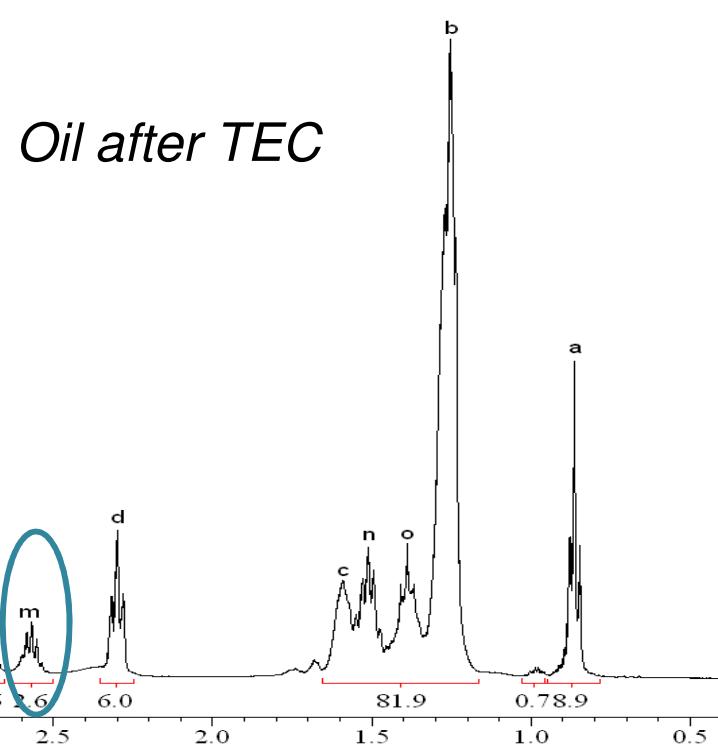
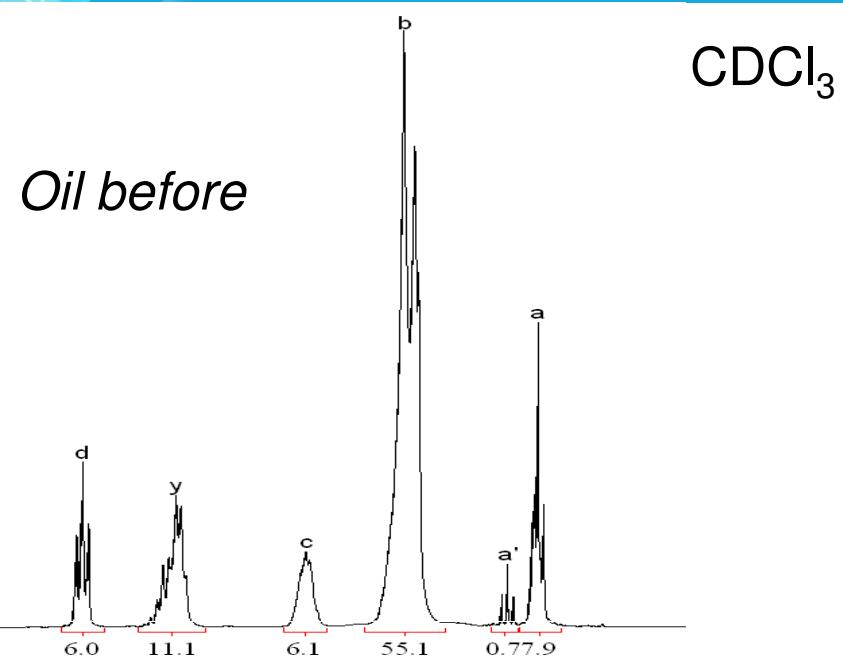
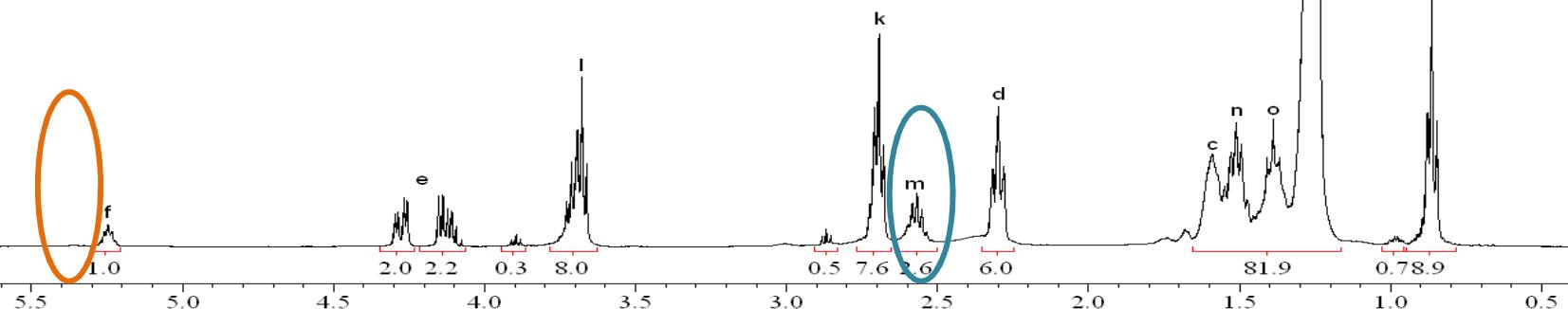
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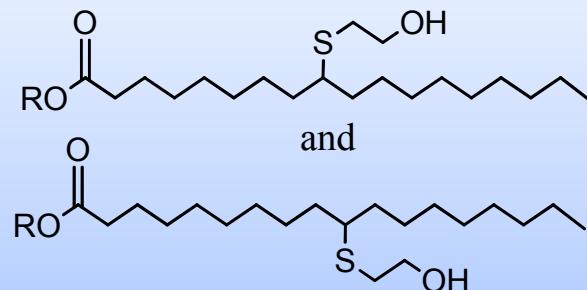


$\text{R}, \text{R}', \text{R}'' =$



Total conversion of $\text{C}=\text{C}$





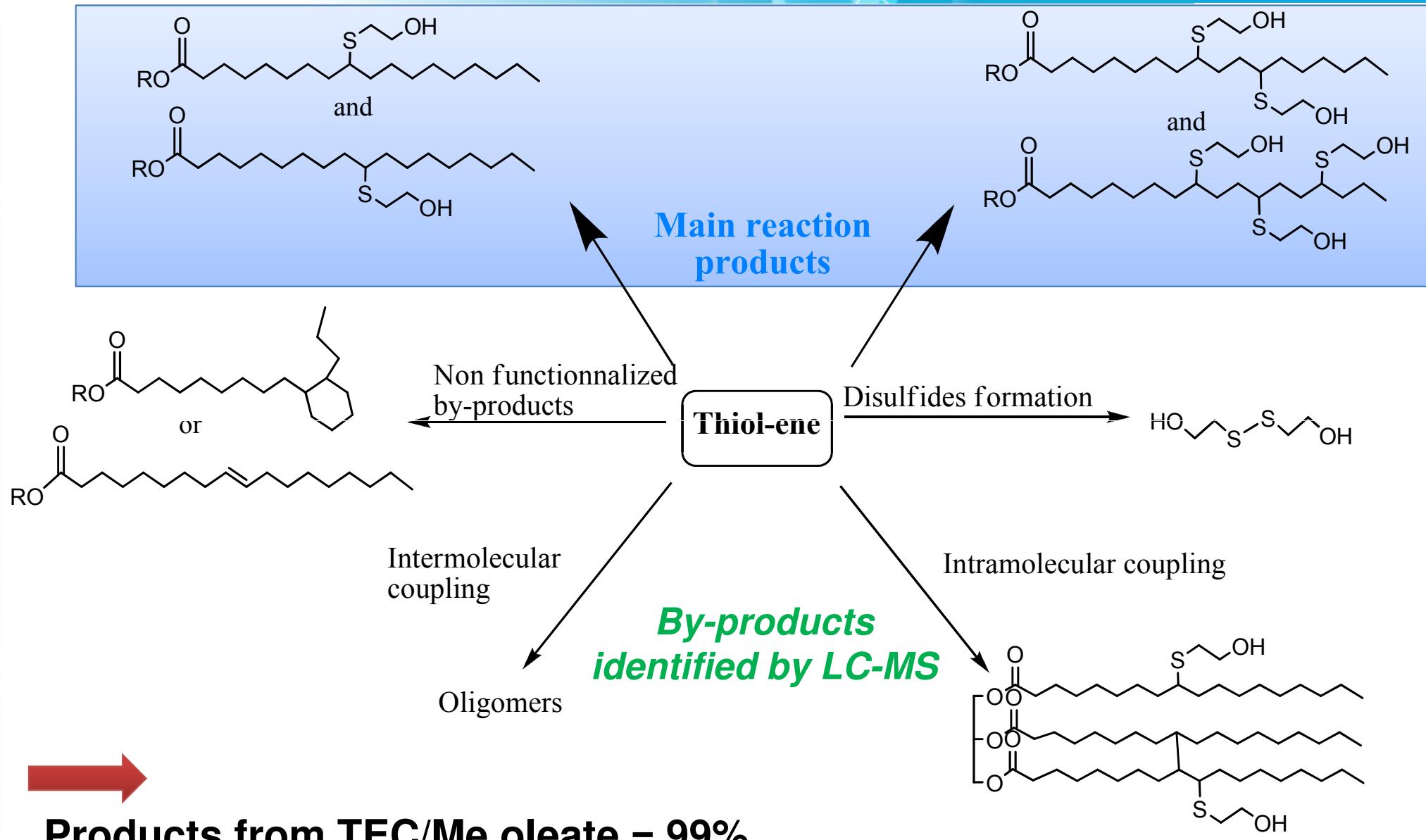
Main reaction products

Thiol-ene



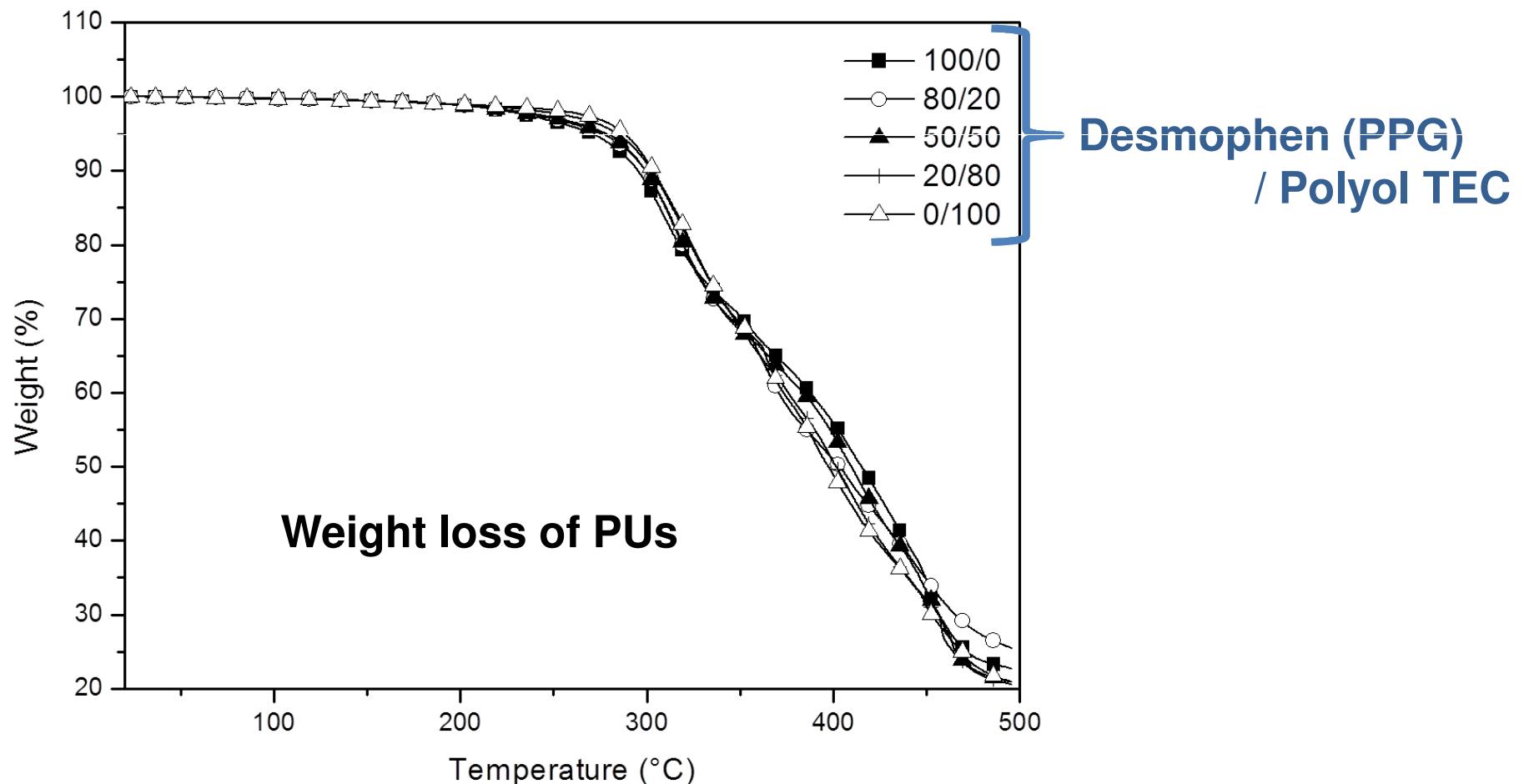
Products from TEC/Me oleate = 99%

The limits to the “click” !

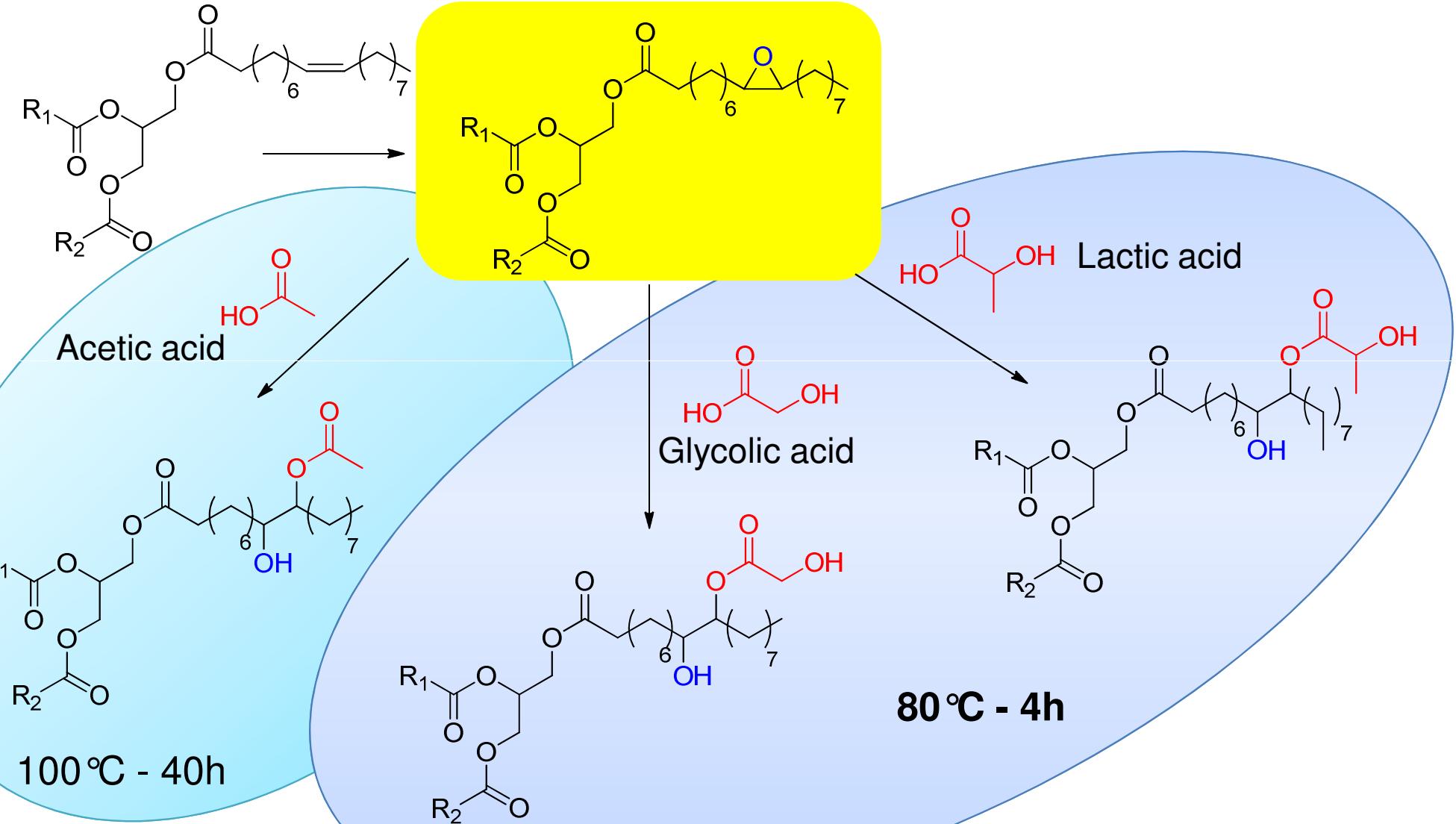


Products from TEC/oil = 65% but functionalized products > 95%

Polyol	Viscosity η_{polyol} (Pa.s)	Eq.wt. (g/eq)	I_{OH} (mg KOH/g)	% OH (%)
Desmophen 1150	0.999	362	165	4.7
Polyol TEC/vegetable Oil	0.999	382	223	4.4

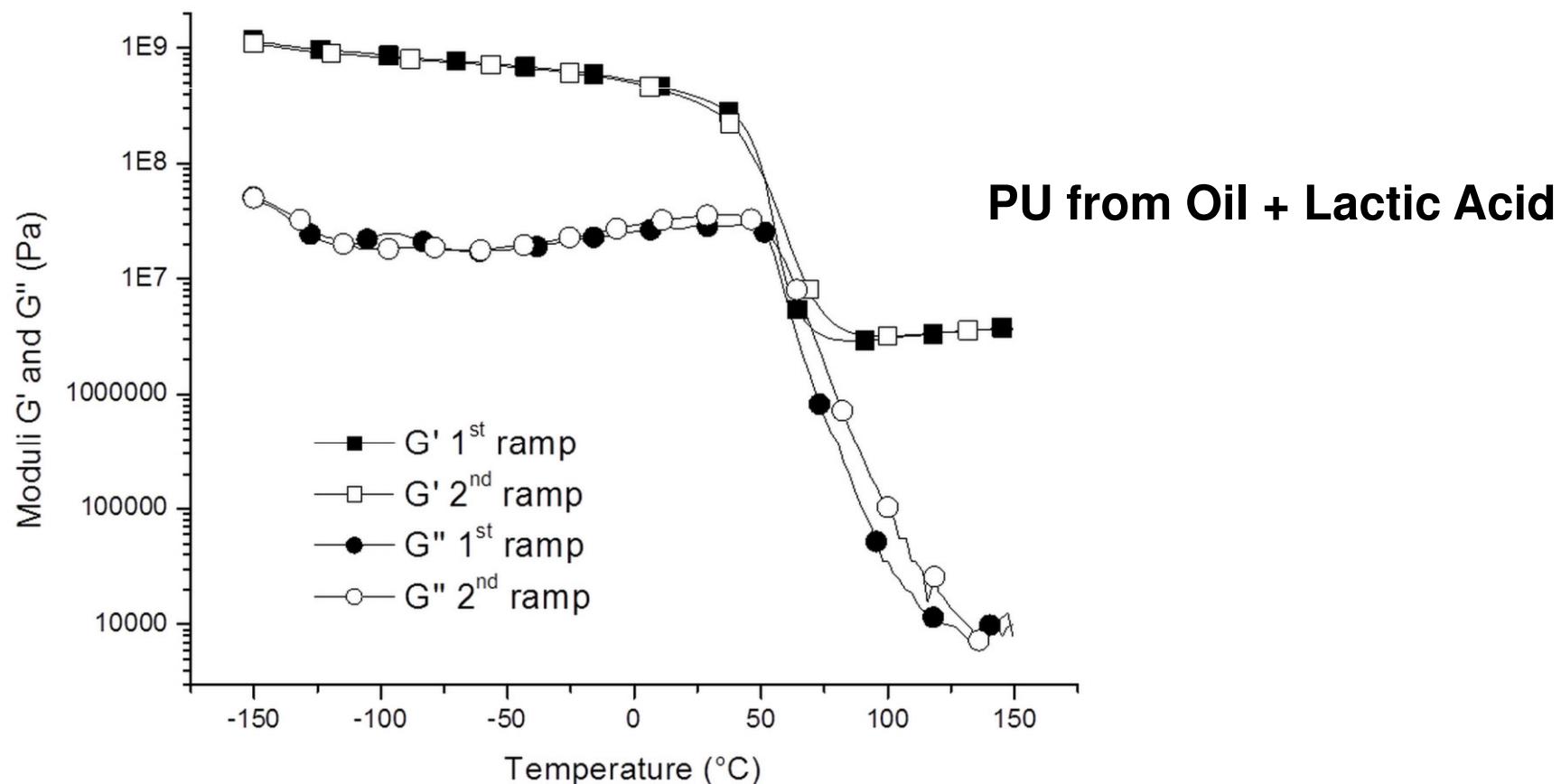


Part 3: New biobased polyols from epoxidized vegetable oils

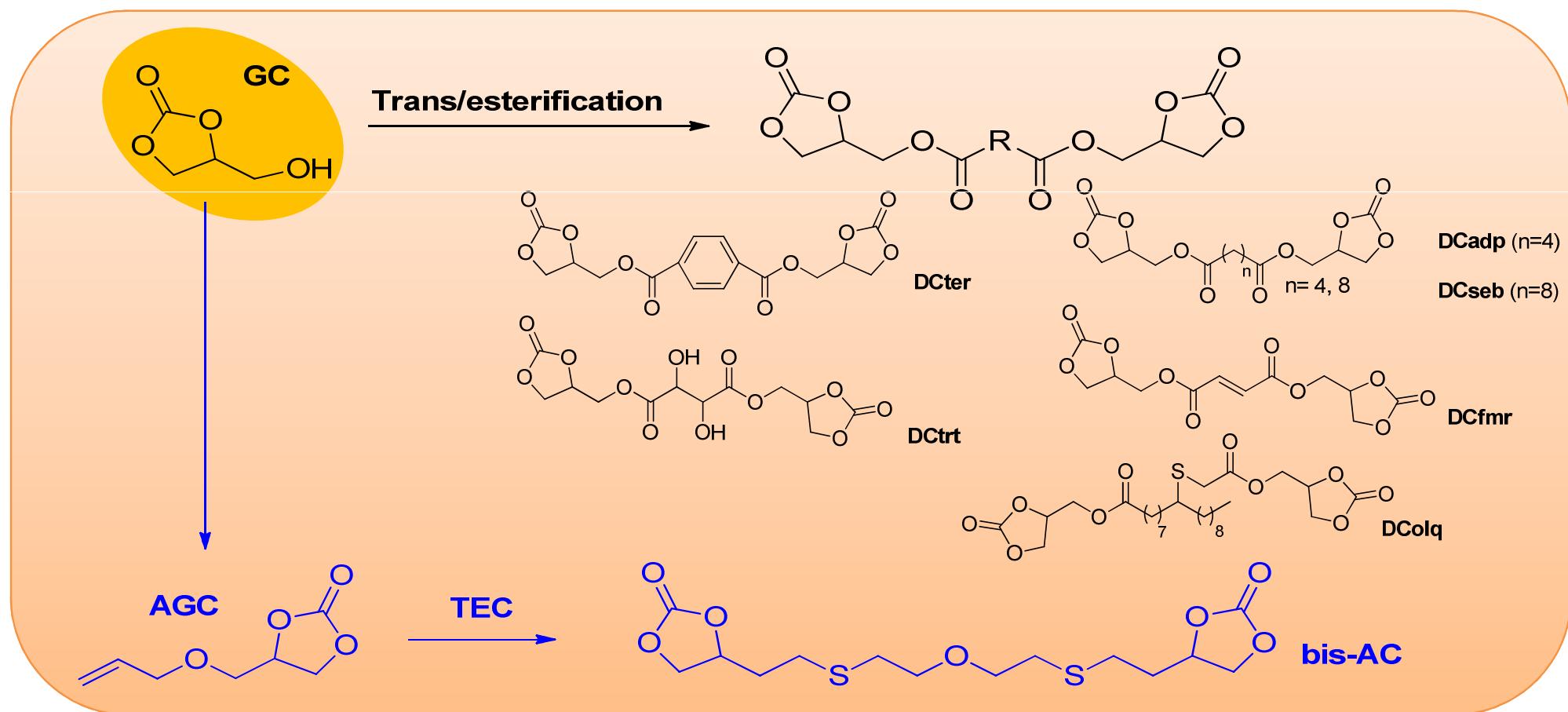
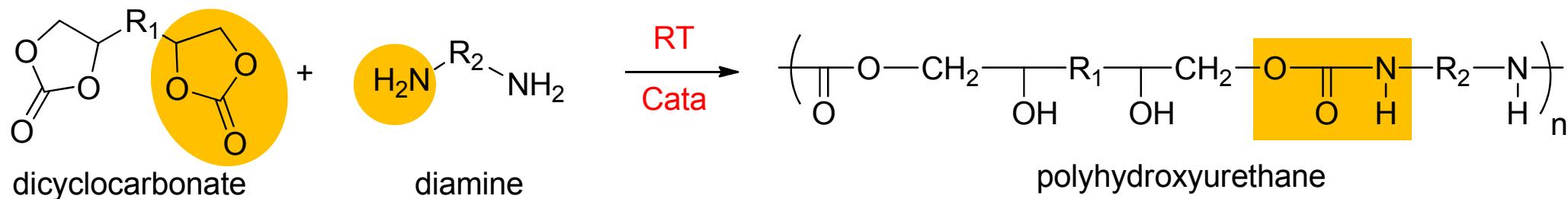


Part 3: New biobased polyols from epoxidized vegetable oils

Polyol	F	Viscosity η_{polyol} (Pa.s)	Eq.wt. (g/eq)	Oligomers (%wt)	I_A (mg KOH/g)
Oil+lactic acid	5.3	47	328	44	3.6
Oil+glycolic acid	4.9	221	276	63	2.6
Oil+acetic acid	4.3	55	298	57	1.8

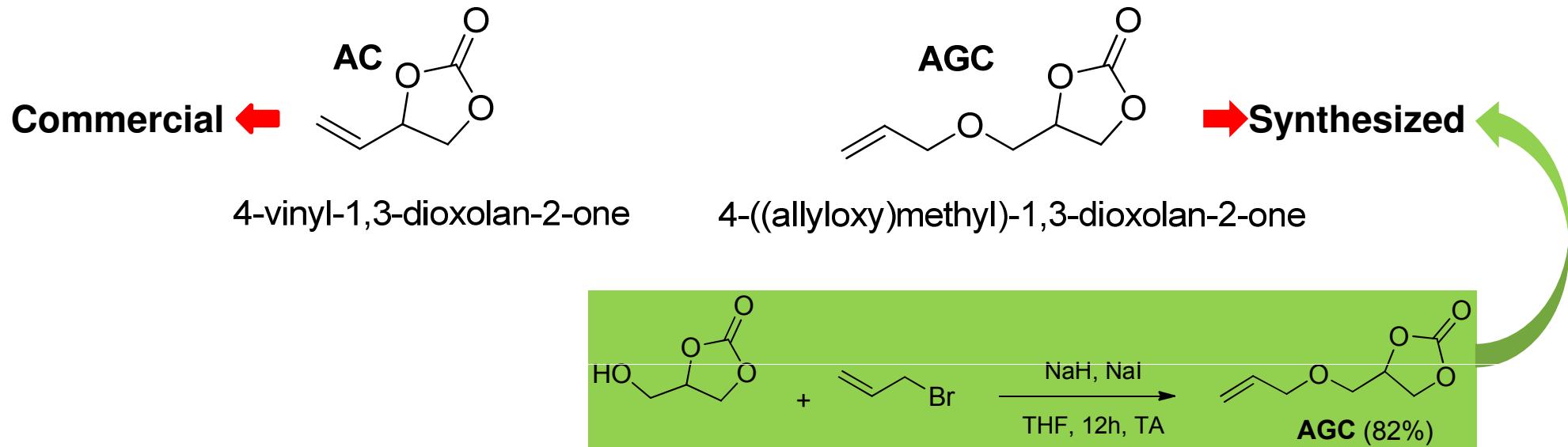


Part 4: Non Isocyanate Polyurethanes

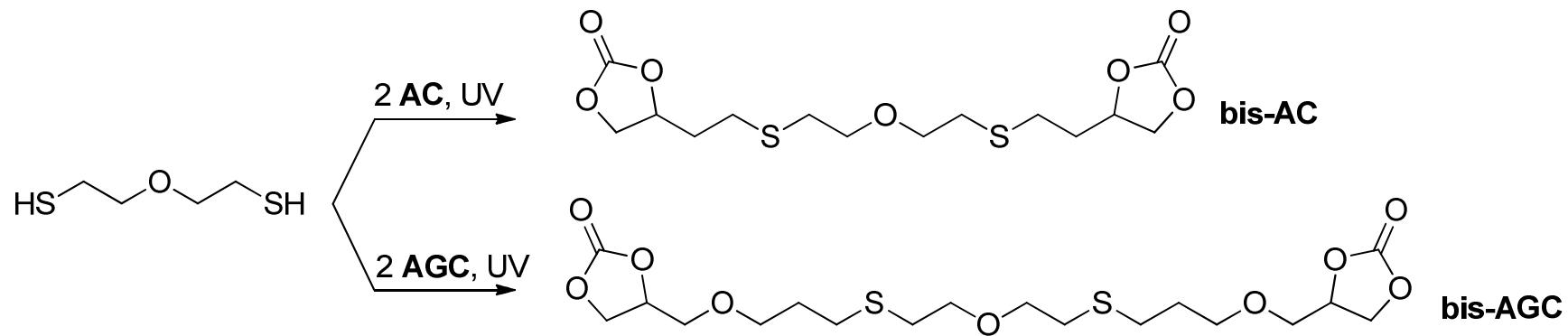


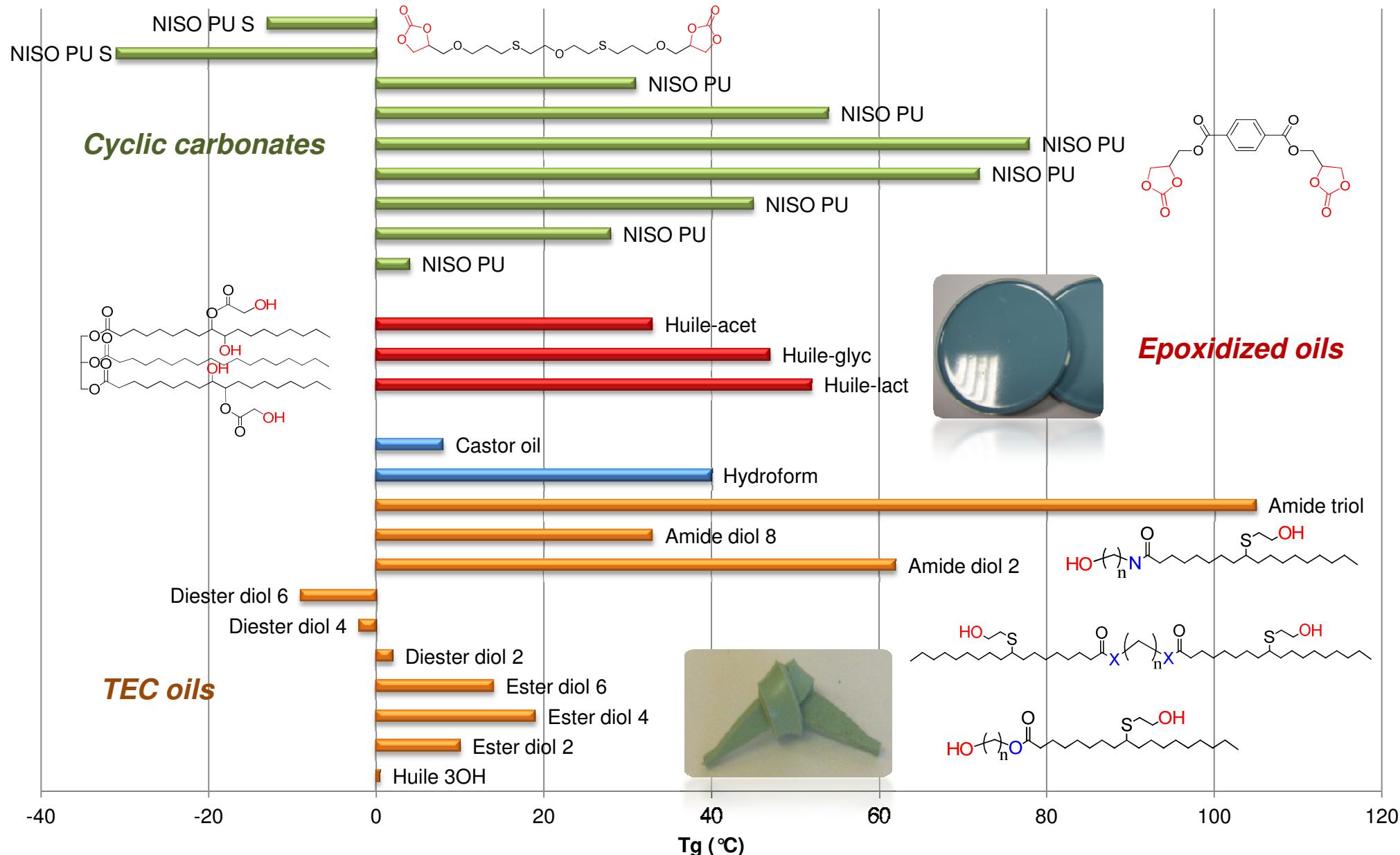
S. Benyahya et al., Polymer Chemistry, 2011, 2, 2661-2667
Polymer International, 2012, Accepted

- Two TEC-cyclocarbonates

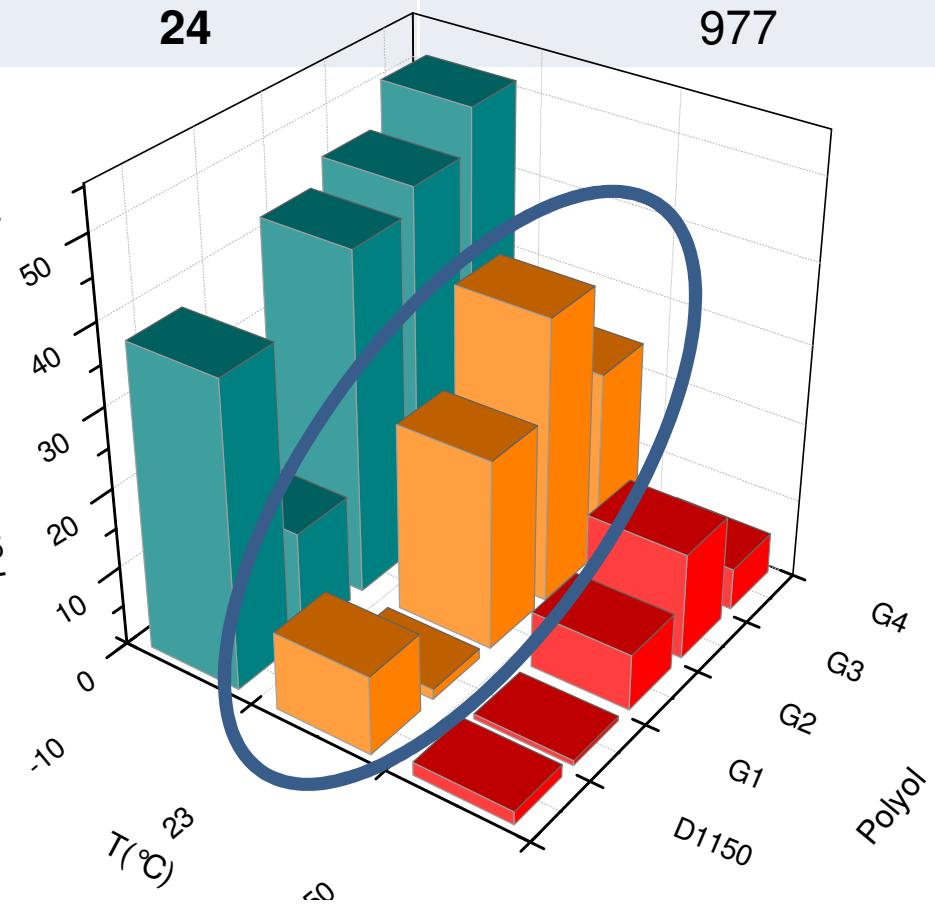


- Synthesis of cyclic carbonates by thiol-ene coupling / UV





T=23 °C	Max elongation (%)	Tensile strength at break (MPa)	Young Modulus N/mm ²
TEC Oil-polyol G1	20	1	7
Polyol-lactate G2	6	24	954
Polyol-glycolate G3	5	36	1284
Polyol-acetate G4	7	24	977



○ Polyol synthesis

- New diols and polyols synthesized easily from vegetable oils and FAME
- Thiol-ene coupling : very efficient reaction, “green” : no solvent, UV without photoinitiator - by-products of TEC used in material synthesis
- New biobased polyols from epoxidized oils – primary or secondary alcohols

○ Cyclocarbonates

- New dicyclocarbonates from glycerin carbonates or by TEC

○ Polyurethanes

- **Biobased** : # 70% of renewable carbon
- Polyols without purification
- **Without isocyanates**, from dicyclocarbonates and amines
- Properties comparable to those from commercial polyols



○ In progress

- « Pilot » scale production of polyols and dicyclocarbonates for PU synthesis by industrial companies partners of project

ACKNOWLEDGEMENT

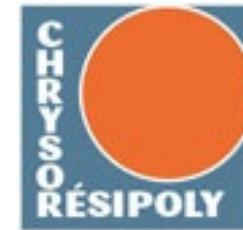
**PROJECT GREENCOAT funded by
ANR MatEtPro**

The Polymer team IAM

Myriam Desroches (PhD)
Dr Sofia Benyahya (Post-doc)
Mylène Stemmelen (PhD)
Dr Vincent Besse (Post-doc)
Dr Rémi Auvergne
Dr Vincent Lapinte
Pr Jean-Pierre Habas
Pr Bernard Boutevin

THE PARTNERS

LCPO



LABORATOIRE DE CHIMIE
DES POLYMIÈRES ORGANIQUES



RÉSIPOLY CHYSOR



SEG DIÉLECTRIQUES

THANK YOU FOR ATTENTION